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Combusted In A Limited Supply Of Pure Oxygen. 0.50g Of The Carbon Combusts To  
Form CO<sub>2</sub> And 0.50 G Of The Carbon Combusts To Form CO. The Resultant Mixture  
Of CO<sub>2</sub> And CO Is Passed Through Excess Na Apr 4th, 2024

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2) Aniline And Acetanilide Stand In A One-to-one Molar Relationship, So 0.0322 Mol  
Of Aniline Produces 0.0322 Mol Of Acetanilide. The Amount Of Acetic Anhydride

Does Not Play A Role, Since It Is In Excess. Then,  $N(N) 0.0322$  .aniline N Aceta II.  
Ide Mol We Find The Molar Mass Of Acetanilide:  $(N)( ) 12 8 1 9 14 1 16 1 135( ) . 89$   
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65 Website Upload 2015 Organic Chemistry Lecture Key Reference Table Q,  
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