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4.  $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$  5.  $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$  Classification Of Chemical Reactions Reactions, Much Like Elements And Compounds, Can Be Classified According To Type. The Ability To Recognize And Classify Reactions Can Help Us Predict 2th, 2024

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026 0026 \_\_ 6 D 20 Al + 02 Al, 0,) + 4 H<sub>2</sub>O + 2D K<sub>2</sub>SO<sub>4</sub> (8) + BaCl<sub>2</sub> + \_\_KCl \_BaSO<sub>4</sub> CsH<sub>12</sub> (D<sub>2</sub> + CO<sub>2</sub> + H<sub>2</sub>O  
Ca (OH) 2 (N) + NHCl (D<sup>2</sup>D<sup>3/4</sup>D<sup>1/2</sup>N<sup>1</sup> N<sup>1</sup> D<sup>1</sup> N<sup>1</sup> N<sup>1</sup> D<sup>2</sup>D<sup>3/4</sup>N<sup>1</sup>) + CaCl NHOHp) VO<sub>5</sub> (4) + H<sub>2</sub>O + Na + D<sup>1/2</sup>; ZnI<sub>2</sub> ... 3th, 2024

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