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Equation Is A Concise Shorthand Expression Which Represents The Relative Amount
Of Reactants And Products Involved In A Chemical Reaction. $\text{CO}_2 + \text{H}_2\text{O} \rightleftharpoons \text{H}_2\text{CO}_3$
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Chemical Reactions For All Combustion Reactions Of Hydrocarbons Or Hydrogen-
carbon-oxygen Molecules, The Products Will Always Be CO_2 And H_2O (assuming A
Complete Reaction). A. Combustion: $\text{C}_6\text{H}_{12} + 9\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$ B.
Combustion: $2\text{C}_4\text{H}_6 + 11\text{O}_2 \rightarrow 8\text{CO}_2 + 6\text{H}_2\text{O}$ C. Combustion: $\text{C}_6\text{H}_{10}\text{O}_3$ Jun
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Predicting Products Of Chemical Reactions - Practice ProblemsName: Suggested
Answers Date: ____ Predicting Products Of Chemical Reactions - Practice Problems
Directions: Predict The Products For, And Then Balance Each Of The Following
Chemical Reactions: 1. $\text{SiI}_4 + \text{Mg}$ (single Replacement) $\text{SiI}_4 + 2\text{Mg} \rightarrow 2\text{MgI}_2 + \text{Si}$ 2. 2Al
 $+ 3\text{I}_2$ (synthesis) $2\text{Al} + 3\text{I}_2 \rightarrow 2\text{AlI}_3$ 3. CuC Jan 5th, 2024Section 3 Predicting The
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