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B Trends In The Periodic Table In Each Group, The Elements Have Similar Properties,  
With A Gradual Trend In Properties Down The Group. This Is Because They Have The  
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Periodic Table Are Metals And Those On The Right Side Are Non-metals. May 22th,  
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Dimitri Mendeleev, A Russian Chemist, Is Credited With Creating The Periodic Table  
Of Elements In 1869. His Breakthrough Came When He Arranged The Elements In  
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Related Groups. Within These Groups, Trends In Common Properties Occur These  
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Table On Periodic Trends, Apr 3th, 2024

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Trends And These Trends Can Be Predicted With The Help Of The Periodic Table. They Can Also Be Explained And Understood By Analyzing The Electron Configurations Of The Elements. This Is Because, Elements Tend To Gain Or Lose Valence Electrons To Jun 2th, 2024

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Lab Activity: Periodic Trends . Introduction: In This Activity, You Will Model Three Properties Of The Elements That Show Trends Across A Period And Down A Group On The Periodic Table. These Properties Are: Atomic Radius: A Measure Of The Size Of One Atom Of An Element . Ionization Energy: The Measure Of The Energy Required To Remove An Electron From The Mar 6th, 2024

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