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5.6 5.8 Kinetic Molecular Theory (KMT) Gas Particles Can Be Considered Point Particles Colliding With The Walls Of An Enclosure Randomly. The Higher The Temperature, The Faster Molecules Move, The More Kinetic Energy They Have. Kinetic Theory Gas Model For KM Mar 3th, 2024

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## Section 1 The Kinetic-Molecular Chapter 10 Theory Of Matter

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#### States Of Matter & The Kinetic Molecular Theory

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1. Describe How Gases, Liquids, And Solids Compare Using The Following Table. Solids Liquids Gases Volume (definite Or Indefinite) Molecular Motion (high, Med, Low) Distance Between Molecules (g 2. Read Page 420: What Is Kinetic Energy? Describe Kinetic Molecular ... Feb 3th, 2024

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Mean Kinetic Energy For A Mole Of Gas • For A Mole Of Ideal Gas, Mean Kinetic Energy, K<sup>-</sup> Is Related To The Root Mean Squared Velocity, Vrms, By L The Mean Kinetic Energy Of A Sample Of Ideal Gas Is Directly Proportional To Absolute Temperature. L The Mean Feb 3th, 2024

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#### CHAPTER 12 GASES AND KINETIC-MOLECULAR THEORY

 Gas.  $\{k1 = K \ 2 \ For \ The \ Same \ Sample \ Of \ A \ Gas \ At \ The \ Same \ T. \ \{Thus \ We \ Can \ Write \ Boyle's \ Law \ Mathematically \ As \ P \ 1V \ 1 = P \ 2V \ 2 \ May \ 2th, \ 2024$ 

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A) Degrees Celsius Are Larger Than Degrees Kelvin. B) 0°K Is Much Warmer Than 0°C. C) Celsius Is A Relative Scale Is Based On The Freezing And Boiling Of Water; Kelvin Is An Absolute Scale. D) There Is No Difference Whatsoever Between Them. 20. What Is The Difference Between The Mar 2th. 2024

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Converting Kelvin To Celsius - 273 Convert 393K To  $C^{\circ}$  - 273 = 20°C Converting Fahrenheit To Celsius (not Used Much) Use C = 5/9(F - 32) Example: Convert 80°F To

°C C = 5/9(F-32) = 26.7°C . Physics 20: Heat Teacher Notes Page 8 Of 24 Practice Problems K = C + 273 C = K - 273 1. Convert Th Feb 3th. 2024

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