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Arranged Elements By Increasing Atomic Mass
Noticed The Repetition Of Properties Every Eighth Element
Created The Law Of Octaves Lothar Meyer (1830–1895)
Demonstrated A Connection Between Atomic Mass And Elements' Properties
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B Trends In The Periodic Table In Each Group, The Elements Have Similar Properties, With A Gradual

Trend In Properties Down The Group. This Is Because They Have The Same Number Of Electrons In The Outer Shells. Elements On The Left Side Of The Periodic Table Are Metals And Those On The Right Side Are Non-metals. 4th, 2024

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