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2. The Apr 3th, 2024

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The Binding Energy Per Nucleon Dictates Results When Nuclides Break Apart (fission) Or Fuse Together (fusion) (keep In Mind That ... (two Protons And Two Neutrons)  ${}^4_2\text{He} + 2 \dots$  Or An Anti-neutrino (  $\bar{\nu}$  ) Neutrinos Have No Charge, Spin  $1/2$ , And Mass  $\sim 0.1 - 1$  eV (?) The Beta Particle Jun 19th, 2024

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32-2 Radioactivity Unstable Nuclei Can Either Decay Into A Stable Nucleus Of Different N And Z, Or Can Return To The Ground State From An Excited State. Three Different Types Of Decay Particles: 1. Alpha Particles, Which Consist Of Two Neutrons And Two Protons, And Are Nuclei Of 2. Electrons Mar 29th, 2024

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Simplest Form, The Form Factor Is The Fourier Transform Of The Charge Density. For Low Momentum Transfer,  $F(q) \gg Z(1 - 1/6 Q^2 R^2 + \dots)$ . (N.B. Actually Scattering Theory And Form Factors Can Be More Complicated Than This, But For The Simplest Cases The Above Are Approximately True.) F( Mar 13th, 2024

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A. A  $^{62}\text{Ni}$  Atom Has More Mass B. The Protons, Neutrons, And Electrons Have More Mass C. They Have Exactly The Same Mass Binding Energy  $V$  The Atomic Mass Of  $^{62}\text{Ni}$  Is 61.928 U. Ni Consists Of 28 Protons, 34 Neutrons, And 28 Electrons. If We

Add Up The Mass Of 28 Proton Jan 28th, 2024

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3 2. This Question Is About Nuclear Reactions. (a) (i) Distinguish Between Fission And Radioactive Decay. (4) A Nucleus Of Uranium-235 ( $^{235}\text{U}$ ) May Absorb A Neutron And Then Undergo Fission To Jun 17th, 2024

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