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Research 794 C. Technological Problem Solving 796
C.1 Technological Problem-Solving Model 796 13th,
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Understandable Format ... A Sterling Silver Ring Has A
Mass Of 12.0 G And Contains 11.1 G Of Pure Silver.
What Is The Percentage Weight By Weight
Concentration Of Silver In The Metal? Solution $M_{Ag} = 11.1 \text{ g}$
 $M_{Alloy} = 12.0 \text{ g}$ $C_{Ag} = \frac{11.1 \text{ g}}{12.0 \text{ g}} \times 100\% = 92.5\%$
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