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2013 U.S. NATIONAL CHEMISTRY OLYMPIAD

A Reaction Mixture Is Prepared Containing 0.60 Mol Of Aluminum And 1.20 Mol Of Manganese Dioxide. The Mixture Is Heated Until One Of The Reactants Has Been Completely Consumed According To The Equation: $2 \text{ Al} + 3 \text{ MnO}_2 \rightarrow 3 \text{ Mn} + \text{Al}_2\text{O}_3$ What Quantity Of Which Reactant Remains Uncombined? (A) 0. 1th, 2024

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Manual Transmission Not Available 5-speed Automatic Transmission CVT 6-speed Fuel Capacity (gallons) 17.2 18.5 Required Fuel Regular Unleaded Regular Unleaded EPA Fuel Economy, City/Highway/Combined (mpg W/automatic Transmission**) 27/36/30 22/31/25 ECON Mode Standard Not Available 2th, 2024

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From: 3/25/2013 Thru: 3/31/2013 Run Date: 4/1/2013 11:30:33 ...

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A Restaurant Would Like To Serve Four Dinner Parties, P1 Through P4. The Restaurant Has A Total Of 8 Plates And 12 Bowls. Assume That Each Group Of Diners Will Stop Eating And Wait For The Waiter To Bring A Requested Item (plate

Or Bowl) To The Table When It Is Required. Assume Th 1th, 2024

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