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(mole-mass, Mass-mole Problems) Stoichiometry Worksheet #2 (mole-mass, Mass-mole Problems) 1. $N_2 + 2O_2 \rightarrow N_2O_4$ A. If 15.0g Of N_2O_4 Was Produced, How Many Moles Of O_2 Were Required? 15.0g N_2O_4 1 Mol N_2O_4 2 Mol O_2 92.0g N_2O_4 1 Mol N_2O_4 = 0.326 Mol O_2 B. If 4.0×10^{-3} Moles Of Oxygen Reacted, How Many Grams Of N_2 Were Needed? 4.0×10^{-3} Mol O_2 1 Mol N_2 28 ... Jul 1th, 2024.

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