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Calculating Molality From Molarity
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3.mole Fraction = 0.950
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3.3 G Of NaCl (58.4 G/mol) And 9 G Of KBr (119.8 G/mol) Were Dissolved In 40 G Of Water (18.0 G/mol). What Is The Mole Fraction Of KBr In The Solution?
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Chemistry 11 Mole Fraction/Molality Worksheet Date
1. A Solution Is Prepared By Mixing 100.0 G Of

Water, H₂O, And 100.0 G Of Ethanol, C₂H₅OH.
Determine The Mole Fractions Of Each Substance. 2.
The Molality Of An Aqueous Solution Of Sugar
(C₁₂H₂₂O₁₁) Is 1.62m. Calculate The Mole Fractions
Of Sugar And Water. 3. Calculate The Molality Of 25.0
Grams Of Feb 1th, 2024K Is The Molal Freezing Point
Depression And M Molality Is ...As 1 Atm = 1.013 ×
10⁵ Pa, The Osmotic Pressure Is $\Pi = (0.0036 \times 1.013$
 $\times 10^5) \text{ Pa} = 365 \text{ Pa}$ Hence $C = \Pi / RT = (365 \text{ Pa}) /$
 $((8.314 \text{ M}^3 \text{ Pa K}^{-1} \text{ Mol}^{-1}) \times (293 \text{ K})) = 0.15 \text{ Mol M}^{-3}$ As
1 M³ = 1000 L, This Corresponds To $C = 1.5 \times 10$ Jun
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