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### **Answer Key**

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Materials Chemistry: Stoichiometry - Problem Sheet 1  
Directions: Solve Each Of The Following Problems.  
Show Your Work, Including Proper Units, To Earn Full  
Credit. 1. Silver And Nitric Acid React According To The  
Following Balanced Equation:  $3 \text{Ag}(s) + 4 \text{HNO}_3(aq) \rightarrow 3 \text{AgNO}_3(aq) + 2 \text{H}_2\text{O}(l) + \text{NO}$  ... Feb 2th, 2024

## **Chemistry Supplemental Stoichiometry Problems**

### **Answer Key**

Chemistry Supplemental Stoichiometry Problems  
Answer Key Stoichiometry Is A Chemical Section  
Involving The Use Of Relationships Between Reagents  
And / Or Products In A Chemical Reaction To  
Determine The Desired Quantitative Data. In Greek,

Stoikhein Means Element And Metricity Means Measurement, Then Steichiometry Literally Translated Means ... Jan 1th, 2024

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Practice Work 53 – Stoichiometry-04 Mixed Stoichiometry Problems General Information You Will Need A Periodic Table, Your Stoichiometry Notes, And Appendix 12 For This Assignment. Sorry About The Lack Of Format. I'm In A Time Crunch. 123.88 G/mol 70.90 G/mol 137.32 ... Jan 2th, 2024

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### **Answer**

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### **Name: KEY**

Chemistry 11 Stoichiometry Review Worksheet

Name: \_\_\_KEY\_\_\_ 33.  $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$

Actual yield = 1.45g Theoretical yield = 5.95g  $\text{KClO}_4 \times 1 \text{ mol}$

$122.55 \text{ g} \times 2 \text{ O}_2 \times 32.00 \text{ g/mol} = 3.11 \text{ g O}_2$

%Yield =  $\frac{1.45 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$  34.  $4\text{FeCO}_3 + \text{O}_2$

$\rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Percent purity = 62.8%? pure  $\text{FeCO}_3$  Jan 1th, 2024

## **Basic Stoichiometry Phet Lab Answer Key**

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Key (ANSWER 386.3g Of  $\text{LiNO}_3$ ) 4) Using The Following

Equation:  $\text{Fe}_2\text{O}_3 + 3\text{H}_2 \rightarrow 2\text{Fe} + 3\text{H}_2\text{O}$ . Calculate

How Many Grams Of Iron Can Be Made From 16.5 Grams Of  $\text{Fe}_2\text{O}_3$  By The Following Equation. Mar 1th, 2024

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