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10 When Vanadium(II) Compounds Are Dissolved In Water, The Following Equilibrium Is Established. $\text{V}^{2+} + \text{H}_2\text{O} \rightleftharpoons \text{V}^{3+} + \text{H}_2\text{O} + \text{OH}^-$ What Would Alter The Composition Of The Equilibrium Mixture In Favour Of The V^{2+} Ions? A Adding An Acid B Adding A Reagent That Selectively Precipitates 2th, 2024

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The Following Data To Calculate A Value For The Lattice Energy Of AgBr(s). First
Ionisation Energy Of Silver = +731kJmol⁻¹ Electron Affinity Of Bromine = 325kJmol⁻¹
Enthalpy Change Of Atomisation Of Silver = +285kJmol⁻¹ Enthalpy Change Of
Atomisation Of Bromine = +112kJmol⁻¹ ... 3th, 2024

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