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### **Ideal Gas Law Worksheet $PV = nRT$**

Ideal Gas Law Worksheet  $PV = nRT$  Use The Ideal Gas Law, " $PV = nRT$ ", And The Universal Gas Constant  $R = 0.0821\text{ L}\cdot\text{atm} / (\text{K}\cdot\text{mol})$  To Solve The Following Problems:  $K\cdot\text{mol}$  If Pressure Is Needed In  $\text{kPa}$  Then Convert By Multiplying By  $101.3\text{ kPa} / 1\text{ atm}$  To Get  $R = 8.31\text{ kPa}\cdot\text{L} / (\text{K}\cdot\text{mole})$  Apr 18th, 2024

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The Purpose Of This Lab Is To Study The Ideal Gas Law To See How The Pressure, Volume, Temperature, And Amount Of A Gas Effect One And Another. Theory The Behavior Of A Gas Depends On A Number Of Variables, Namely Pressure, P, Volume, V, Temperature, T, And The Amount Of Gas, N. These Variables Are Related To Each Other By An Equation Of State ... Mar 21th, 2024

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