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Worksheet 7 - Ideal Gas Law I. Ideal Gas Law The Findings Of 19th Century Chemists And Physicists, Among Them Avogadro, Gay-Lussac, Boyle And Charles, Are Summarized In The Ideal Gas Law:  $PV = nRT$   $P$  = Pressure  $V$  = Volume  $n$  = Moles Of Gas,  $R$  = Universal Gas Constant  $T$  = Temperature. The Value Of  $R$  Varies With The 1th, 2024

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A Perfect Radiator ("black Body") Would Have  $E=1$ . A Perfect Reflector ("shiny" Object) Would Not Radiate At All;  $E=0$ . Lecture 33 10/28 Radiation This Behavior Is Contained In The Stefan-Boltzmann Law: Here,  $E$  Is 3th, 2024

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## Raoult's Law - Ideal Solution (□□□□□□□□□□□□□□)

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## Ideal Gas Law Problems - Mmsphyschem.com

Ideal Gas Law Problems 1) How Many Molecules Are There In 985 ML Of Nitrogen At 0.0° C And 1.00 X 10<sup>-6</sup> Mm Hg? 2) Calculate The Mass Of 15.0 L Of NH<sub>3</sub> At 27° C And 900. Mm Hg. 3) An Empty Flask Has A Mass Of 47.392 G And 47.816 G When Filled With Acetone 2th, 2024

## Ideal Gas Law Worksheet $PV = nRT$

Ideal Gas Law Worksheet PV = NRT Use The Ideal Gas Law, "PerV-nRT", And The Universal Gas Constant R = 0.0821 L\*atm To Solve The Following Problems: K\*mol If Pressure Is Needed In KPa Then Convert By Multiplying By 101.3kPa / 1atm To Get R =8.31 KPa\*L / (K\*mole) 4th, 2024

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### **Exp 8 Ideal Gas Law - HCC Learning Web**

Due Before Lab Begins. Answer In The Space Provided.

1. A) Determine The Number Of Moles Of Krypton Gas Contained In A 3.25 Liter Tank At A Pressure Of 5.80 Atm And A Temperature Of 25.5 °C. B) If The Gas Was Oxygen Instead Of Krypton, Will The Answer Be The Same? Why Or Why Not? 2. A) 0.190 G Of A Gas Occupies 250.0 ML At STP. What Is Its ... 1th, 2024

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