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## **Electron Arrangement Of The Electron Configuration Of Atoms**

The Distance From The Bottom Of The Chart Indicates The Energy Of Each Energy Level And Sub-level. The Closer The Energy Level Is To The Bottom Of The Chart, The Lower Its Energy. At The Bottom Of The Chart, You Will find The first Energy Level,  $N = 1$ . The Chart Shows Only One Circle In The first Energy Level. 1th, 2024

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At The Orbital Filling Diagrams Of The Atoms Listed In Table 5.3. An Oxygen Atom Contains Eight Electrons. The Orbital Of Lowest Energy, 1s, Has One Electron, Then A Second Electron Of Opposite Spin. The Next Orbital To Fill Is 2s. It Also Has One Electron, Then A Second Electron Of Opposite Spin. 4th, 2024

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