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Interacting With The Solvent. The Ethoxide Ion Is Not As Sterically Hindered So It Can Accommodate More Solvent Interactions. As A Result Ethoxide Is Better Solvated And Is Therefore More Stable Than Tert-butoxide (Figure 3.7). Jul 2th, 2024  
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- Draw An H-bond Between An Ether And An Alcohol.
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- Without The Crown Ether, The Solubility Of KF In Benzene ...
- Consider A Retrosynthetic Disconnect On The T-butyl Side.
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