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Section 3 Classifying Chemical Reactions  
Decomposition Reaction- One Substance Breaks Down, Or Decomposes, Into Two Or More Substances.  $AB \rightarrow A + B$  Most Decomposition Reactions Require The Use Of Heat, Light, Or Electricity.  
1th, 2024  
Classifying Chemical Reactions  
Types Of Chemical Reactions: 1. Synthesis 2. Decomposition 3. Combustion 4. Single-Replacement 5. Double-replacement \*Classifying How Atoms Rearrange During A Chemical Reaction. December 12, 2014  
1. Synthesis Reaction: 2 Or More Substances React To Produce A Single Product  $A + B \rightarrow AB$   
2th, 2024  
Classifying Chemical Reactions Worksheet  
Classifying Chemical Reactions Worksheet Name: \_\_\_\_\_ Period: \_\_\_\_\_  
Classify Each Reaction As Acid/base, Redox, Synthesis, Decomposition, Single Replacement, Double Replacement Or Combustion. They May Be More Than One. The Equations Are Not Balanced. 1. 1th, 2024.  
Section 8-3: Classifying Chemical Reactions  
Section 8-3: Classifying Chemical Reactions  
Reaction Type General Format Characteristics Of Reactants Characteristics Of Products  
Example (s) Decomposition  $AB \rightarrow A + B$  Only One Reactant Two Or More Simpler Products  
 $CuCO_3 (s) \rightarrow CuO + CO_2 (g)$   
 $2 H_2 O (l) \rightarrow 2 H_2 (g) + O_2 (g)$   
Synthesis AKA: Combination  $O (l) + B \rightarrow AB$   
2 Two Elements Or Two Simple 3th, 2024  
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