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1001 Page© Www.CHEMSHEETS.co.uk 23-Feb-2016 Chemsheets A2 1001 Page 3 As The Temperature Changes, The Average Energy Of The Particles Increases And So The 23th, 2024© Www.CHEMSHEETS.co.uk 2-July-2016 Chemsheets A2 1076 Page 1A) [Co(H₂O)₆]²⁺ Co +2 E) [FeO₄]²⁻ Fe +6 I) K₂[CoCl₄] Co +2 B) [CrCl₆]³⁻ Cr

+3 F) $[\text{Mn}(\text{CN})_6]^{4-}$ Mn +2 J) $\text{K}_3[\text{AuF}_6]$ Au +3 C) $[\text{Co}(\text{NH}_3)_6]\text{Cl}_2$ Co +2 G) $[\text{Ni}(\text{CO})_4]$ Ni 0 K) $(\text{NH}_4)_2[\text{IrCl}_6]$ Ir +4 D) $[\text{Co}(\text{NH}_3)_5\text{Cl}]\text{Cl}_2$ Co +3 H) $[\text{Ni}(\text{EDTA})]^{2-}$ Ni +2 L) $\text{Na}[\text{Mn}(\text{CO})_5]$ Mn -1 TASK 2 - Writing Half Equations A) $2\text{I}^- \rightarrow \text{I}_2 + 2\text{e}^-$ B ... 5th, 2024.

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1Metal Carbonate \rightarrow Metal Oxide + Carbon Dioxide (on Heating) $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$ TASK 4 - WRITING BALANCED EQUATIONS 1) Balance The Following Equations.

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