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Chemistry: Stoichiometry - Problem Sheet 2 KEY 9) 2.24×10^{23} 2.22×10^{23} 4.63×10^{23} molecules | 1 Mol | 6.02×10^{23} molecules | 1 Mol Cl | 1mol 71 G Cl | 54.6 g Cl | 10.292 g Ag | 1 Mol Ag | 108 g Ag | 1 Mol Cu | 1 Mol Ag | 63.5 g Cu File Size: 108KB Page Count: 4 Apr 2th, 2024

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Chemistry 11 Stoichiometry Review Worksheet Name: KEY 33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$ Actual yield = 1.45g
Theoretical yield = $5.95\text{g KClO}_4 \times \frac{1\text{mol}}{122.55\text{g}} \times 2 \times \frac{32.00\text{g}}{1\text{mol}} = 3.11\text{g O}_2$ %Yield = $\frac{1.45\text{g}}{3.11\text{g}} \times 100\% = 46.6\%$ 34.
 $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$ Percent purity = 62.8%? g pure FeCO_3 Mar 2th, 2024

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