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5. List The Seven Diatomic Molecules. Chapter 9: Chemical Names & Formulas 1. Determine Whether The Following Compounds Are Ionic Or Covalent, And Then Name Them. CaSO 4, N 2 O 4, NH 4 NO 3, CaS, Li 3 PO 4, NH 3, Cu(OH) 2. 2. Complete The Following Table And Name Each Ionic Compound Formed. + May 1th, 2024

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C. Ag-107, Au-197 And Ar-40 11. List The Order That You Fill The Energy Levels And Orbitals For Electron Configuration. 12. Fill In The Orbital Diagram Below For Tungsten And Then Write The Full Electron Configuration Below. W: 13. Which Of The Following Is An Incorrect Orbital Diagram? Why? Jan 1th, 2024

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2 Chapter 2 – Atomic Theory Subatomic Particles: • Proton (+) Found In Nucleus; The Nucleus Was Discovered By Rutherford In The Gold Foil Experiment • Neutron (neutral); Found In Nucleus; Heaviest Subatomic Particle • Electron (-) Found Outside The Nucleus; Lightest Subatomic Particle; 1800 Electrons = 1 Proton Jan 1th, 2024

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NAME FORMULA Ionic/Acid/Molecular 1. Carbon Tetrachloride 2. Magnesium Sulfate 3. Iron (II) Oxide 4. Zinc Chloride 5. Copper (I) Phosphite 6. Aluminum Oxide 7. Carbonic Acid 8. Phosphorus Tribromide 9. Ammonium Hydroxide 10. Tin (IV) Carbonate 11. Pb 3 (PO 4) 2 12. Hg(CN) 2 13. AgNO 3 14. Na Jan 1th, 2024

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