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1-carbon 2-oxygen Thus One Mole Of CO<sub>2</sub> Contains  
Three Times Avogadro's # Of Atoms. Example: How  
Many Atoms Are In 2.12 Mole Of Propane (C<sub>3</sub>H<sub>8</sub>)?  
Know: # Of Moles = 2.12 Mole C<sub>3</sub>H<sub>8</sub> 1 Mole C<sub>3</sub>H<sub>8</sub>  
= 6.02 X 10<sup>23</sup> Molecules C<sub>3</sub>H<sub>8</sub> 1 Molecule C<sub>3</sub>H<sub>8</sub>  
= 11 Atoms ( 3 Carbon Atoms & 8 Hydrogen Atoms) 1th,  
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