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Chapter 3 Stoichiometry

Chapter 3 Stoichiometry 3-7 EXAMPLE PROBLEM: Determine The Molar Mass Of A Compound Calculate The Molar Mass For Each Of The Following Compounds: (a) 2-Propanol, $\text{CH}_3\text{CH}(\text{OH})\text{CH}_3$ (b) Iron(II) Phosphate, $\text{Fe}_3(\text{PO}_4)_2$ SOLUTION: You Are Asked To Calculate The Molar Mass Of A Compound. You Are Given The Compound's Formula. Jan 3th, 2024

CHAPTER 2: Atoms, Molecules And Stoichiometry

2.5 Stoichiometry And Equations Stoichiometry 1) Stoichiometry Is The Proportion Of Things Either Reacting Or Combining. 2) In Compounds, It Refers To The Ratio In Which The Atoms Are Combined Together. For Example, Water, H_2O Has A Stoichiometry Of 2 Hydrogen To 1 Oxygen. 3) It Also Refers To The Reacting Proportions In A Chemical Equation. Apr 3th, 2024

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