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### **CELLULAR RESPIRATION: Cellular Respiration Equation ...**

CELLULAR RESPIRATION: • Cellular Respiration Equation (Products And Reactants)  $C_6H_{12}O_6 + O_2 \rightarrow CO_2 + H_2O + ENERGY$  REACTANTS PRODUCTS • Oxidation/Reduction (include Examples) O Oxidation: Lose Electrons (LEO) Ex. Glucose, NADH, FADH<sub>2</sub> Are OXIDIZED O Reduction: Gain 4th, 2024

### **Cellular Respiration Pre-Reading Cellular Respiration Pre ...**

Cellular Respiration Pre-Reading Questions Use The Reading And Diagram On The Bottom Flip To Complete This Page. 1. Where Does Photosynthesis Occur? \_\_\_\_\_  
\_\_\_\_\_ 2. Where Does Cellular Respiration Occur? \_\_\_\_\_  
\_\_\_\_\_ 3. Glucose Is Another Name For \_\_\_\_\_ 4. Photosynthesis And Cellular Respiration 4th, 2024

## **Unit 4: Cellular Respiration Notes Cellular Respiration Is ...**

Unit 4: Cellular Respiration Notes Cellular Respiration Is The Process By Which Food Is Broken Down By The Body's Cells To Produce Energy In The Form Of ATP Molecules. A. Cellular Respiration Overview: 1. Cellular Respiration Is Carried Out By Every Cell In Both 3th, 2024

### **Respiration 1 Cellular Respiration**

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UNIT 5: CELLULAR CHEMISTRY (CELLULAR RESPIRATION) Big Idea: ENERGY Biological Systems Use Energy And Molecular Building Blocks To Grow, Reproduce, And Maintain Homeostasis. 5. For Learning Target #5, Construct A Venn Diagram That Shows The Similarities And Differences Between Cellular Respiration And Photosynthesis... 3th, 2024

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Fermentation Enables Cells To Produce ATP Without Oxygen Fermentation Is A Way Of Harvesting Chemical Energy That Does Not Require Oxygen. Fermentation Takes Advantage Of Glycolysis, Produces Two ATP Molecules Per Glucose, And Reduces  $\text{NAD}^+$  To  $\text{NADH}$ . The Trick Of Fermentation Is To Provide An Anaerobic Path For Recycling  $\text{NADH}$  Back To  $\text{NAD}^+$ . 3th, 2024

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Enzyme Will Return To Its Original Active Form. Figure 5.4 Shows The Relationship Of The Structure Of Fumaric Acid To Succinic Acid. Experiment: In This Experiment, A Hydrogen Acceptor That Does Not Occur In Nature, The Dye Methylene Blue, Is Used. When Oxidized It Is Blue, But When It Accepts Hydrogen And Is Thus Reduced, It Becomes 1th, 2024

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1. Which Bonds Represent High Energy E-? 2. Describe The Change In Potential Energy Of The E-3. Name The Molecules That Carry The E-to The ETC  $O_2 + H^+$  4. Which Part Of The Diagram Represents Oxidative Phosphorylation? 4th, 2024

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