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And The Neutron That Is Formed Have A Mass Number Of 1. As With Positron Emission, However, The Atomic Number Of The D Jan 4th, 2024
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$2\text{MnO}_2 + \text{CO}_2 \rightarrow \text{ClO}_2 + \text{ClO}_2 + \text{ClO}_3$
 $\text{Cu}(\text{NH}_3)_4^{2+} + \text{S}_2\text{O}_4^{2-} \rightarrow \text{SO}_3^{2-} + \text{Cu} + \text{NH}_3$
 $\text{Zn} + \text{NO}_3^- \rightarrow \text{Zn}(\text{OH})_4^{2-} + \text{NH}_3$
 $\text{Al} + \text{OH}^- \rightarrow \text{AlO}_2^- + \text{H}_2$
 $\text{Zn} \rightarrow \text{Zn}(\text{OH})_4^{2-} + \text{H}_2$

Answers . Acidic: $14\text{H}^+ + 2\text{Mn}^{2+} + 5\text{BiO}_3^- \rightarrow 2\text{Mn}$

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