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C. Nuclear Reactions Often Produce Large Amounts Of Energy Because Small Amounts Of Mass Are Converted Into Energy (see Einstein's Famous Equation, E=mc2) D. All Radioactive Isotopes Decay Completely And Disappear Within A Short Time (1 Year Or Less) 30. SrPredict The Decay Pathway For 90. (Strontium-88 Is The Most Abundant Stable Isotope For ... 24th, 2024

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Chemical Reactions In Which Energy Is Released Are Exothermic. The Energy That Is Released Was Originally Stored In The Chemical Bonds Of The Reactants. Often The Heat Given

Off Causes The Product(s) To Feel Hot. Any Reaction That Involves Combustion (burning) Is An Exothermic Chemical Reaction. 2th, 2024

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Section 3 Classifying Chemical Reactions

Decomposition Decomposition Reaction- One Substance Breaks Down, Or Decomposes, Into Two Or More Substances. AB \rightarrow A + B Most Decomposition Reactions Require The Use Of Heat, Light, Or Electricity. 15th, 2024

Classifying Chemical Reactions

Classifying Chemical Reactions Types Of Chemical Reactions: 1. Synthesis 2. Decomposition 3. Combustion 4. Single-Replacement 5. Double-replacement *Classifying How Atoms Rearrange During A Chemical Reaction. December 12, 2014 1. Synthesis Reaction: 2 Or More Substances React To Produce A Single Product A + B AB 2th, 2024

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Section 8-3: Classifying Chemical Reactions

Section 8-3: Classifying Chemical Reactions Reaction Type General Format Characteristics Of Reactants Characteristics Of Products Example (s) Decomposition AB A + B Only One Reactant Two Or More Simpler Products CuCO 3 (s) CuO + CO 2 (g) 2 H 2 O (l) 2 H 2 (g) + O 2 (g) Synthesis AKA: Combination O (l) A + B AB 2 Two Elements Or Two Simple 9th, 2024

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