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Is Formed During The Complete Combustion Of
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$Q = mc\Delta T$. $Q = 50 \times 4.2 \times 11.0$. $Q = 2310 \text{ J}$. To Convert J Into KJ Divide By 1000. $Q = 2.3 \text{ (kJ)}$ Tip: Many Candidates Use The Mass Of A Solid Reactant, In This Case 15.22g, As The Mass In The $Q = mc \Delta T$ Equation. This Is Incorrect! The Energy Taken I 2th, 2024

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